Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Kaushik Banerjee,^a‡ Axel Patrick Ligon,^a Markus Schürmann,^b Hans Preut^b* and Michael Spiteller^a

^aInstitut für Umweltforschung, Universität Dortmund, Otto-Hahn-Str. 6, 44221 Dortmund, Germany, and ^bFachbereich Chemie, Universität Dortmund, Otto-Hahn-Str. 6, 44221 Dortmund, Germany

‡ Current/postal address: National Research Centre for Grapes, Pune 412 307, India.

Correspondence e-mail: uch002@uxp1.hrz.uni-dortmund.de

Key indicators

Single-crystal X-ray study T = 291 K Mean σ (C–C) = 0.004 Å Disorder in main residue R factor = 0.034 wR factor = 0.069 Data-to-parameter ratio = 12.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved

Methyl (*Z*,*Z*)-*α*-(methoxyimino)-2-[({1-[3-(trifluoromethyl)phenyl]ethylidene}amino)oxymethyl]benzeneacetate

The crystal structure of the title compound, $C_{20}H_{19}F_3N_2O_4$, contains one molecule in the asymmetric unit. The CF₃ group is disordered. Important features of the molecule are the two C—N bonds in a *Z*,*Z* configuration, one in the β -methoxy-stem and the other in the oxymethyl side-chain between the two aromatic rings.

Received 26 February 2004 Accepted 2 March 2004 Online 13 March 2004

Comment

The title compound, (I), is an isomer of the fungicide trifloxystrobin (TFS). TFS appears in an E,E configuration (Ebeling *et al.*, 2003) but, on exposure to the environment, it is susceptible to conversion to a mixture of four isomers. As part of our investigations of the isomerization of TFS in the environment, we have determined the crystal structure of (I). The general synthesis route of TFS was described by Ziegler *et al.* (2003).

Experimental

Compound (I) was produced by illuminating a solution of TFS (200 mg) in acetone (50 ml) in a Suntest apparatus at 500 W m^{-2} in a quartz vessel. Compound (I) appeared as a component of a mixture of isomers of TFS and was identified by comparison with the chromatogram of the reference standard obtained from Bayer Crop Science, Germany. Compound (I) was isolated from the mixture of isomers by preparative high-pressure liquid chromatography (HPLC), employing a Gynkotek HPLC equipped with a preparative Kromasil 100 C18 column (7 $\mu m,\,250\times20$ mm) and a UV detector set at 250 nm. The mobile phase was acetonitrile-water (90:10 v/v) at a rate 4 ml min^{-1} . Compound (I) eluted with a good baseline separation at a retention time of 21.81 min. The fraction corresponding to (I) was collected separately and evaporated to complete dryness at 303 K under vacuum. The purity of the white solid thus obtained was confirmed by IR, Raman, NMR and mass spectrometry. It was redissolved in a minimum quantity of methanol and kept inside a fume-hood at room temperature for slow evaporation. Compound (I) crystallized out over a period of a week.

Crystal data

 $C_{20}H_{19}F_3N_2O_4$ $M_r = 408.37$ Monoclinic, $P2_1/n$ a = 8.0122 (11) Åb = 6.9993 (6) Å c = 36.417 (4) Å $\beta = 91.945 (5)^{\circ}$ V = 2041.1 (4) Å³ Z = 4

Data collection

Nonius KappaCCD diffractometer	$R_{\rm int} = 0.038$
ω scans	$\theta_{\rm max} = 25.1^{\circ}$
16 494 measured reflections	$h = -9 \rightarrow 9$
3538 independent reflections	$k = -8 \rightarrow 8$
1275 reflections with $I > 2\sigma(I)$	$l = -43 \rightarrow 4$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.034$	$w = \exp[4.20(\sin\theta/\lambda)^2)]/[\sigma^2(F_o^2)]$
$wR(F^2) = 0.069$	$(\Delta/\sigma)_{\rm max} = 0.001$
S = 1.01	$\Delta \rho_{\rm max} = 0.09 \ {\rm e} \ {\rm \AA}^{-3}$
3538 reflections	$\Delta \rho_{\rm min} = -0.13 \text{ e } \text{\AA}^{-3}$
291 parameters	

 $D_x = 1.329 \text{ Mg m}^{-3}$

Cell parameters from 16 494

Mo $K\alpha$ radiation

reflections

 $\mu = 0.11 \text{ mm}^{-1}$

T = 291 (1) K

_{max} = 25.1° $= -9 \rightarrow 9$ $= -8 \rightarrow 8$ $= -43 \rightarrow 43$

Block, colourless

 $0.20 \times 0.15 \times 0.15 \mbox{ mm}$

 $\theta = 3.0-25.1^{\circ}$

H atoms were placed in calculated positions, with C-H = 0.93-0.97 Å, and were treated as riding, with $U_{iso} = 1.5U_{eq}(C)$ for methyl groups and $1.2U_{eq}(C)$ for others; methyl groups C9 and C19 were allowed to rotate but not to tip, whereas methyl group C10 was positioned purely geometrically. Each F atom is disordered over two positions with an occupancy of 0.5.

Data collection: COLLECT (Nonius, 1998); cell refinement: HKL DENZO and SCALEPACK (Otwinowski & Minor, 1997); data reduction: HKL DENZO and SCALEPACK; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL-Plus (Sheldrick, 1991); software used to prepare material for publication: SHELXL97, PARST95 (Nardelli, 1995) and PLATON (Spek, 2001).

The authors thank the Ministry of Science and Technology, Government of India, for awarding a BOYSCAST fellowship to KB. Thanks are also due to Bayer Crop Science, Germany,

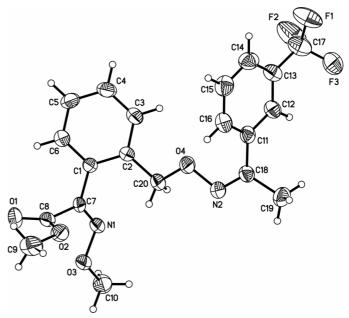


Figure 1

The molecular structure of (I), showing the labelling of all non-H atoms. Displacement ellipsoids are shown at the 30% probability level. The F atoms are disordered, and for each F atom, only one of the two split positions is shown.

for providing the reference substance, technical suggestions and literature support.

References

- Ebeling, M., Heimann, K.-G., Schoefer, S. & Sonder, K. (2003). Pflanzenschutz-Nachr. Bayer, 56, 231-245.
- Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
- Nonius (1998). COLLECT. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). Methods in Enzymology, Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307-326. New York: Academic Press.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1991). SHELXTL-Plus. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.

Spek, A. L. (2001). PLATON. University of Utrecht, The Netherlands.

Ziegler, H. J., Benet-Buchholz, W. Etzel & Gayer, H. (2003). Pflanzenschutz-Nachr. Bayer, 56, 213-230.